



Sp24HWSet2

Homework Problem Set 2

Iverson CH320N

Due Monday January 29

NAME (Print): \_\_\_\_\_

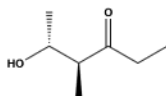
SIGNATURE: \_\_\_\_\_

**Chemistry 320N  
Dr. Brent Iverson  
2nd Homework  
January 23, 2024**

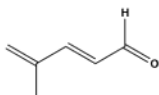
**Please print the  
first three letters  
of your last name  
in the three boxes**

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(3 pts each) Write an accurate IUPAC name for the following molecules.

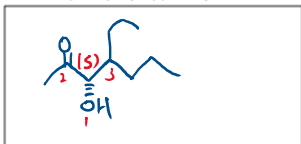


(4S, 5R)-5-hydroxy-4-methyl-3-hexanal  
*methylhexan-3-one*

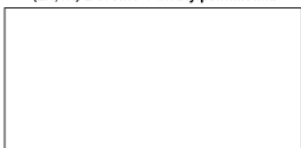


(3 pts each) Draw the correct structure for the given IUPAC name. Use wedges and dashes to show the appropriate stereochemistry where appropriate.

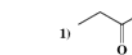
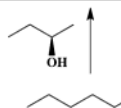
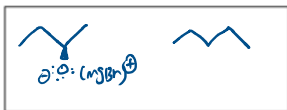
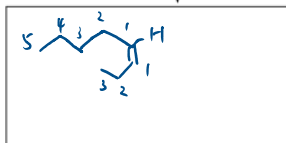
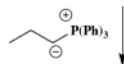
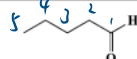
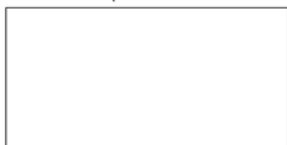
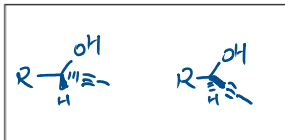
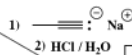
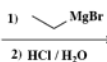
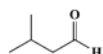
(S)-3-hydroxy-4-propyl-2-heptanone



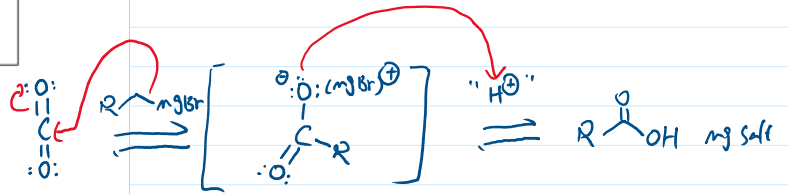
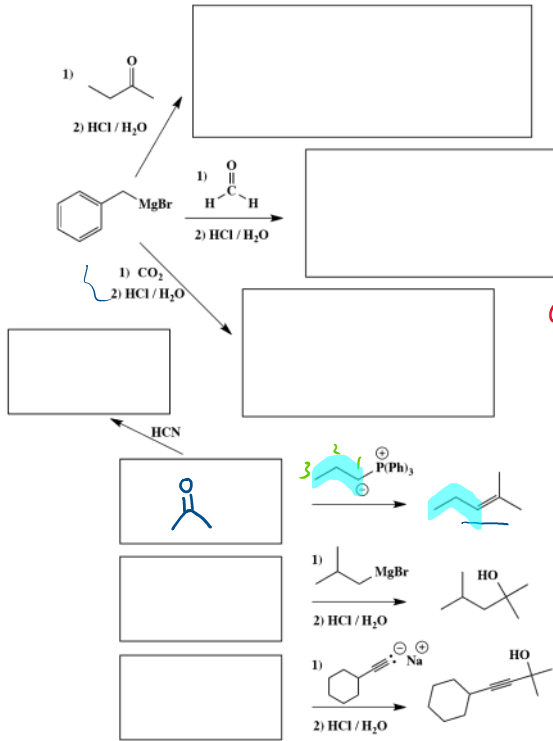
(2R,4R)-2-bromo-4-methylpentanal



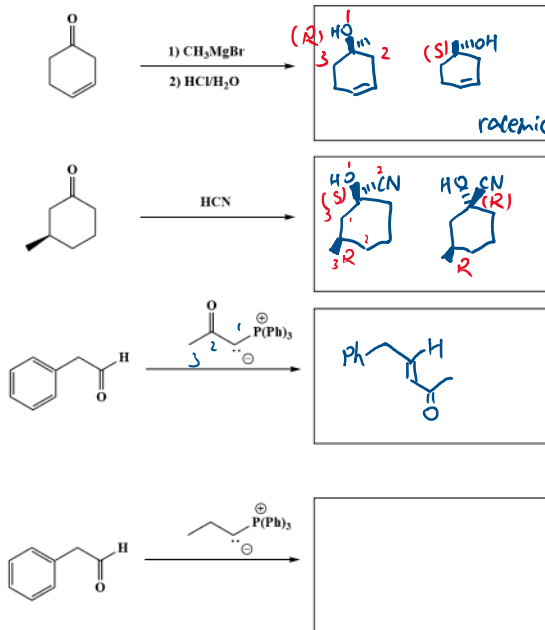
(3 or 5 pts each) Fill in the boxes with the structures that complete the reactions. Use wedges and dashes to indicate stereochemistry when appropriate. If a racemic mixture is formed, you must draw both enantiomers and write "racemic" next to the two structures.



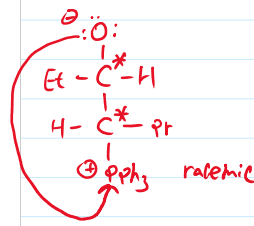
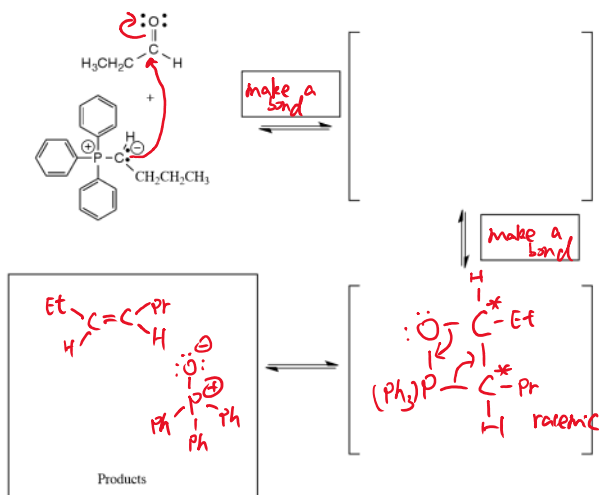
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(18 pts. total) Complete the mechanism for the following Wittig reaction. **Be sure to show arrows to indicate movement of all electrons, write all lone pairs, all formal charges, and all the products for each step.** Remember, I said all the products for each step. **IF A RACEMIC MIXTURE IS CREATED IN AN INTERMEDIATE, MARK ALL CHIRAL CENTERS WITH AN ASTERISK AND WRITE RACEMIC.** IF A RACEMIC MIXTURE IS CREATED IN THE FINAL PRODUCTS, YOU NEED TO DRAW BOTH ENANTIOMERS, AND WRITE RACEMIC. I realize these directions are complex, so please read them again to make sure you know what we want.

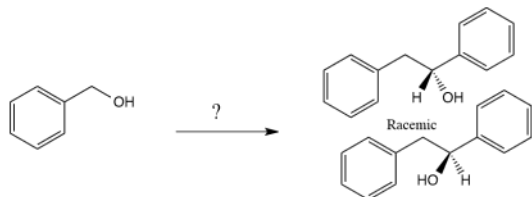


2 pts In the boxes provided adjacent to the first two sets of arrows, write which of the four basic mechanistic elements are involved (i.e. "Make a bond", "Add a proton", etc).

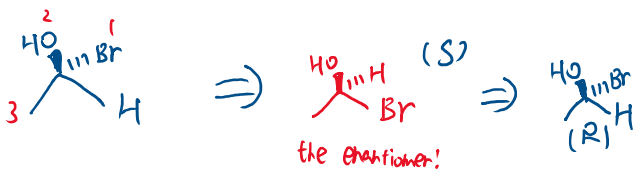
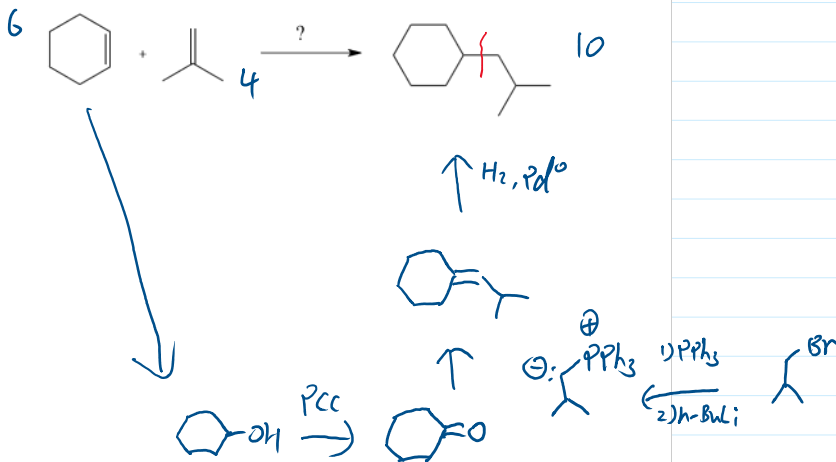
NOTICE THIS

These are synthesis questions. You need to show how the starting material can be converted into the product(s) shown. You may use any reactions we have learned. Show all the reagents you need. Show each molecule synthesized along the way and be sure to pay attention to the regiochemistry and stereochemistry preferences for each reaction.

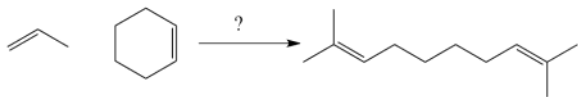
(10 pts) All of the carbon atoms of the products must come from the starting material for this one!



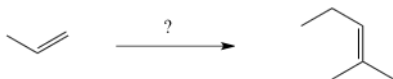
(12 pts) All of the carbon atoms of the products must come from the starting materials for this one!



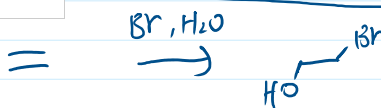
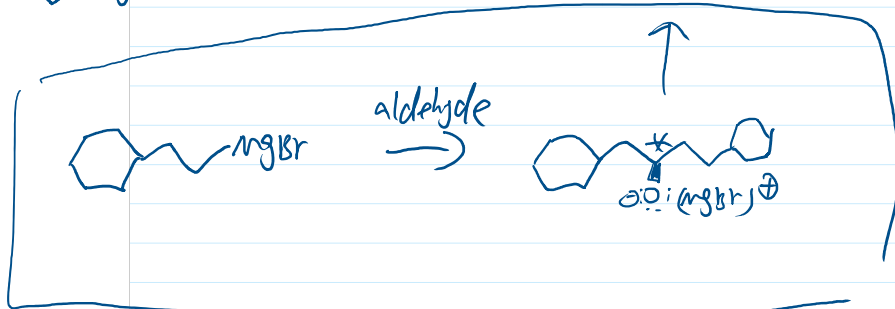
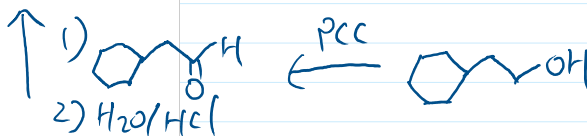
(12 pts) All of the carbon atoms of the products must come from the starting material for this one!



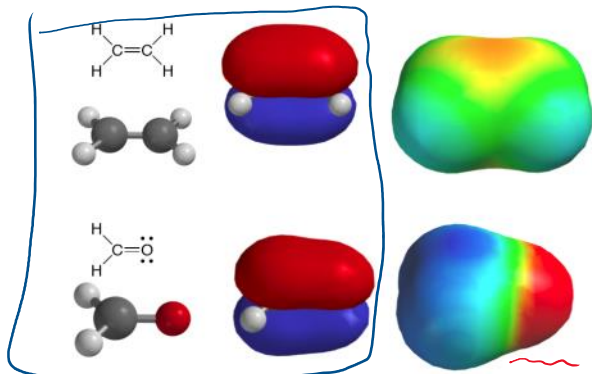
(15 pts) All of the carbon atoms of the products must come from the starting material for this one!



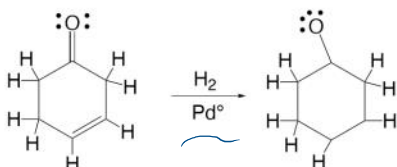
2017 midterm Q16 P15



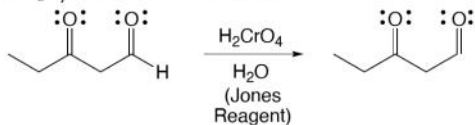
Detour: Hydrogenation and Oxidation of Aldehydes and Ketones



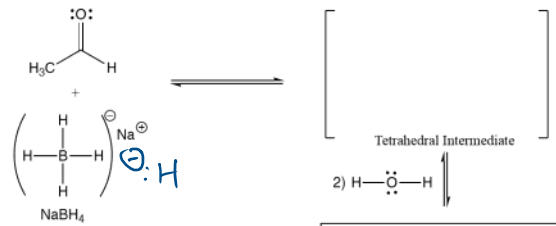
The pi bonds of carbonyls react the same as pi bonds of alkenes with  $H_2$  in the presence of  $Pt^0$ ,  $Pd^0$  or  $Ni^0$



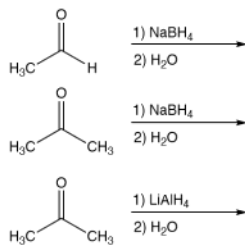
Aldehydes are oxidized to carboxylic acids using the Jones Reagent ( $H_2CrO_4$  in  $H_2O$ ). Ketones do not react.



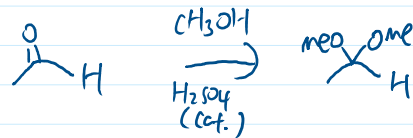
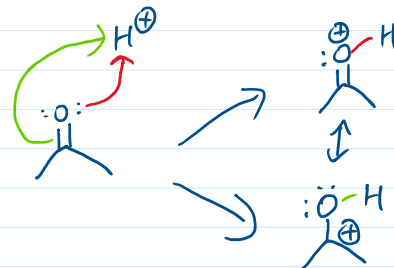
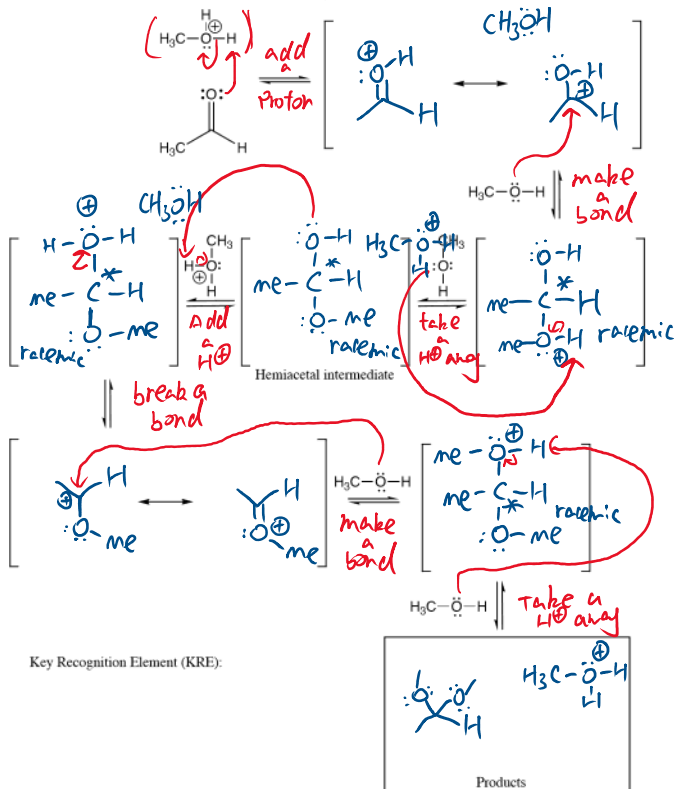
Sodium Borohydride Reacting with an Aldehyde or Ketone



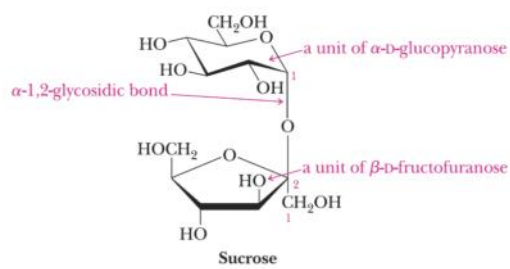
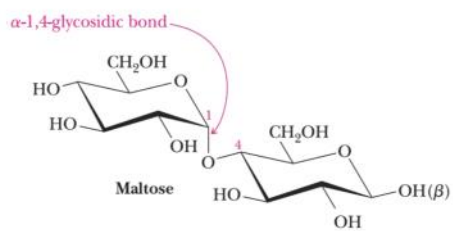
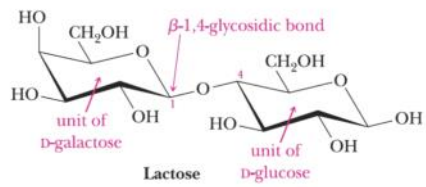
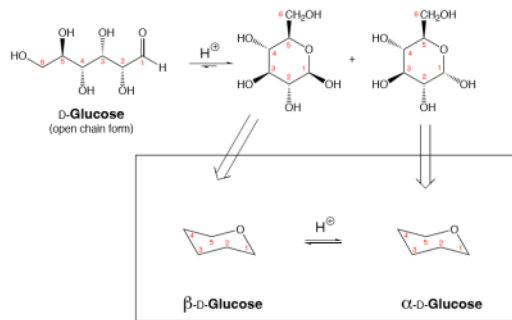
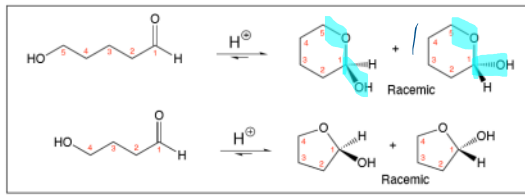
Key Recognition Element (KRE):



Acid Catalyzed Hemiacetal and Acetal Formation From an Aldehyde or Ketone

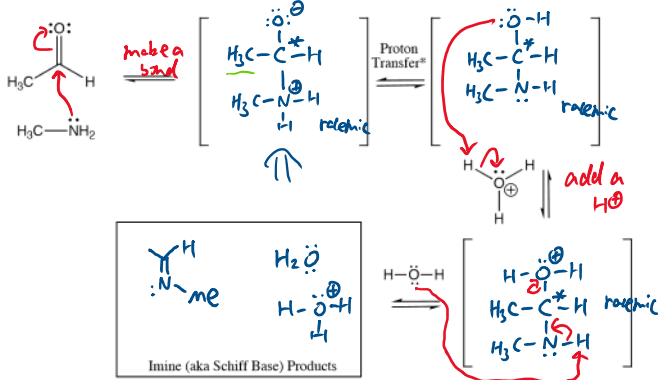


## Cyclic Hemiacetals and Carbohydrates



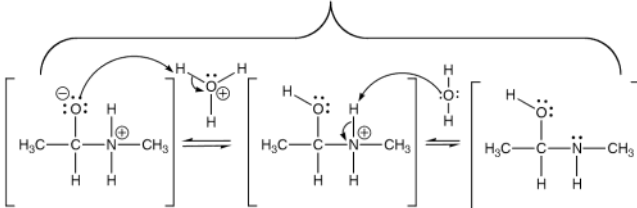


Formation of an Imine (Schiff Base) From an Aldehyde or Ketone Reacting with an Amine

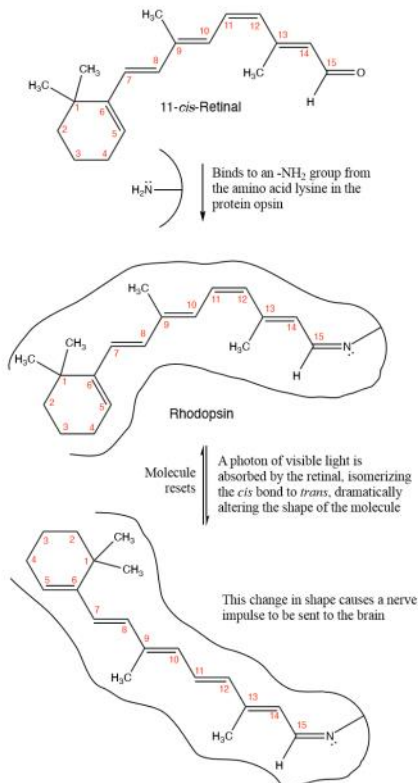


\* "Proton Transfer" refers to a situation in which a proton moves from one part of a molecule to another on the SAME MOLECULE. We do not draw arrows for proton transfer steps because that would be deceptive. In some cases, the same proton may move from one part of the molecule to the other directly, but in other cases, solvent molecules may be involved as indicated in the following scheme. To make things even more interesting, the following two steps might even be reversed in some cases. Because of all the ambiguity, we just write "Proton Transfer" and do not bother with arrows.

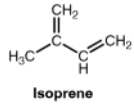
"Proton Transfer"



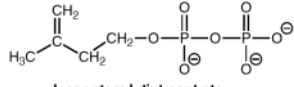
How vision works



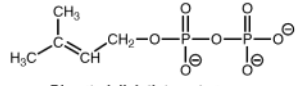
# Terpenes



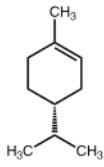
Isoprene



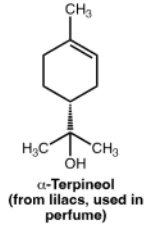
Isopentanyl diphosphate



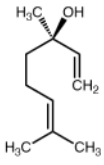
Dimethylallyl diphosphate



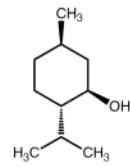
Limonene  
(citrus flavor)



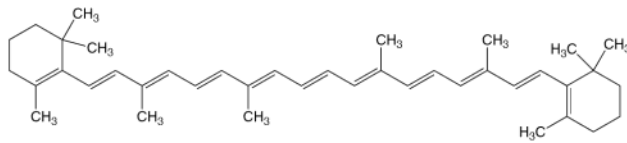
$\alpha$ -Terpineol  
(from lilacs, used in perfume)



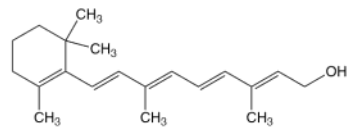
(R)-(-)-Linalool  
(from lavender, used in perfume)



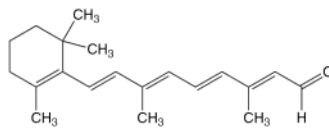
(-)-Menthol  
(common flavoring from peppermint)



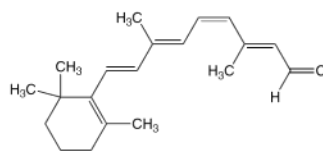
$\beta$ -Carotene



Vitamin A



All trans Retinal



11-cis-Retinal